1 **Plutonium isotopic signatures in soils and their variation (2011-2014) in**
2 **sediment transiting a coastal river in the Fukushima Prefecture, Japan**
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4 Hugo Jaegler¹, Fabien Pointurier², Yuichi Onda³, Amélie Hubert², J. Patrick Laceby^{1,4},
5 Maëva Cirella² & Olivier Evrard^{1*}
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7 **Affiliations**

8 ¹Laboratoire des Sciences du Climat et de l'Environnement, LSCE/IPSL, Unité Mixte de Recherche 8212 (CEA-
9 CNRS-UVSQ), Université Paris-Saclay, F-91198 Gif-sur-Yvette, France

10 ²CEA, DAM, DIF, F-91297 Arpajon, France

11 ³Center for Research in Isotopes and Environmental Dynamics (CRIED), University of Tsukuba, Tsukuba,
12 Japan

13 ⁴Environmental Monitoring and Science Division, Alberta Environment and Parks, 3115 – 12 Street NE
14 Calgary, Alberta, Canada

15 (*) Corresponding author (email address: olivier.evrard@lsce.ipsl.fr).

16 Graphical abstract

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18 Abstract

19 The Fukushima Daiichi Nuclear Power Plant (FDNPP) accident resulted in a significant release of
 20 radionuclides that were deposited on soils in Northeastern Japan. Plutonium was detected at trace levels
 21 in soils and sediments collected around the FDNPP. However, little is known regarding the spatial-
 22 temporal variation of plutonium in sediment transiting rivers in the region. In this study, plutonium
 23 isotopic compositions were first measured in soils (n=5) in order to investigate the initial plutonium
 24 deposition. Then, plutonium isotopic compositions were measured on flood sediment deposits (n=12)
 25 collected after major typhoon events in 2011, 2013 and 2014. After a thorough radiochemical
 26 purification, isotopic ratios ($^{240}\text{Pu}/^{239}\text{Pu}$, $^{241}\text{Pu}/^{239}\text{Pu}$ and $^{242}\text{Pu}/^{239}\text{Pu}$) were measured with a Multi-
 27 Collector Inductively Coupled Mass Spectrometer (MC ICP-MS), providing discrimination between
 28 plutonium derived from global fallout, from atmospheric nuclear weapon tests, and plutonium derived
 29 from the FDNPP accident. Results demonstrate that soils with the most Fukushima-derived plutonium
 30 were in the main radiocaesium plume and that there was a variable mixture of plutonium sources in the

flood sediment samples. Plutonium concentrations and isotopic ratios generally decreased between 2011 and 2014, reflecting the progressive erosion and transport of contaminated sediment in this coastal river during flood events. Exceptions to this general trend were attributed to the occurrence of decontamination works or the remobilisation of contaminated material during typhoons. The different plutonium concentrations and isotopic ratios obtained on three aliquots of a single sample suggest that the Fukushima-derived plutonium was likely borne by discrete plutonium-containing particles. In the future, these particles should be isolated and further characterized in order to better understand the fate of this long-lived radionuclide in the environment.

Capsule

Plutonium concentrations and isotopic ratios in sediment transiting a river draining the main Fukushima radioactive pollution plume decreased with time.

Keywords

FDNPP accident; river lag deposits; Multi Collector-Inductively Coupled Plasma-Mass Spectrometer; Actinide; Pu atom ratios; Source identification

Highlights

- Measurement of Pu in soil and river sediment by MC ICP-MS.
- Determination of Pu sources by isotopic composition.
- FDNPP-derived Pu deposition was heterogeneous in the environment.
- Pu contribution from FDNPP accident varied between 1.1–39.2% of total plutonium.
- Variability of concentrations suggests the occurrence of discrete Pu-bearing particles.

Introduction

The 2011 Fukushima Dai-ichi Nuclear Power Plant (FDNPP) accident released significant quantities of radioactive contamination into the environment. Although ~80% of the atmospheric fallout from the accident occurred over the ocean, the remainder was deposited on Japanese soils and formed a major radioactive pollution plume extending ~70 km northwest of the FDNPP (Evrard et al., 2015; Kawamura et al., 2011). Most of the research conducted after the accident has focused on gaseous radionuclides or radiocaesium isotopes (Bailly du Bois et al., 2012; Estournel et al., 2012; TEPCO, 2012). Other radionuclides, such as plutonium isotopes were also studied, although to a lesser extent (Steinhauser, 2014).

Plutonium is present in the Northern Hemisphere environment as a result of nuclear weapon tests ([Yamamoto et al., 2014](#)). ^{239}Pu ($T_{1/2}=24,110$ y, alpha-decay) and ^{240}Pu ($T_{1/2}=6,563$ y, alpha-decay) are the most abundant isotopes in the environment on a global scale, while ^{238}Pu ($T_{1/2}=88.74$ y, alpha-decay), ^{241}Pu ($T_{1/2}=14.35$ y, beta-decay) and ^{242}Pu ($T_{1/2}=376,000$ y, alpha-decay) are present in smaller concentrations ([Hardy et al., 1973](#); [Harley, 1979](#); [Kelley et al., 1999](#); [Yang et al., 2015](#)). As the isotopic composition of plutonium is related to its origin, these isotopic ratios provide a powerful tool to discriminate between different sources of plutonium in the environment ([Muramatsu et al., 2003](#)). Specific isotopic signatures have been reported for the global fallout associated with nuclear weapon testing, local nuclear weapon tests ([Chiappini et al., 1999](#)), the Chernobyl accident ([Ketterer et al., 2004](#)) and the FDNPP accident ([Evrard et al., 2014b](#)).

Plutonium concentrations associated with the global fallout were shown to be heterogeneous in different aliquots ([Kelley et al., 1999](#)), due to the particulate form of deposition ([Salbu, 2011](#)). Contamination from the global fallout and the FDNPP have very different origins and their concentrations may change differently with time: global fallout contamination originated from the stratosphere, with many episodes spread over several decades, whereas FDNPP was supplied by a single local source at a much lower altitude, during a few days only. In contrast, plutonium isotopic ratios associated with the global fallout are homogeneous ([Kelley et al., 1999](#)) and $^{241}\text{Pu}/^{239}\text{Pu}$ isotopic ratios may be used in particular to calculate the respective plutonium concentrations originating from the FDNPP accident or the global fallout. Due to the relative short half-life of ^{241}Pu (14.35 y), the global fallout source of this isotope has decayed significantly, whereas plutonium released by the FDNPP accident contains abundant levels of ^{241}Pu (1.1×10^{11} - 2.6×10^{11} Bq, ([Zheng et al., 2012b](#))). Its detection in the environment therefore provides unambiguous evidence of the FDNPP release. ([Evrard et al., 2014b](#)).

The plutonium isotopic composition in the fuel from the FDNPP was reconstructed at the moment of the accident based on simulations ([Kirchner et al., 2012](#); [Nishihara et al., 2012](#); [Schwantes et al., 2012](#)) (Table 1). The global fallout atom ratios were also quantified ([Kelley et al., 1999](#); [Muramatsu et al., 2003](#); [Yang et al., 2015](#); [Zhang et al., 2010](#)) (Table 1). The fate of FDNPP-derived plutonium can therefore be investigated in the environment based on $^{240}\text{Pu}/^{239}\text{Pu}$, $^{241}\text{Pu}/^{239}\text{Pu}$ and $^{242}\text{Pu}/^{239}\text{Pu}$ isotopic ratios as demonstrated by Evrard *et al.* ([2014b](#)).

Zheng *et al.* ([2013](#)) published a comprehensive review of plutonium releases from the FDNPP accident. Since then, additional measurements reported by various authors on litter, soil, and river water samples collected in the Fukushima Prefecture after the accident indicated that some environmental samples

contained plutonium from the FDNPP reactors ([Kimura et al., 2015](#); [Utsunomiya et al., 2012](#); [Xu et al., 2016](#)) (Table S1, Figure 1). However, these studies were mainly based on measurements by alpha-spectrometry of the $^{239+240}\text{Pu}$ and ^{238}Pu activities and of the $^{238}\text{Pu}/^{239+240}\text{Pu}$ isotopic ratio. Using alpha-spectrometry, FDNPP-derived plutonium was detected only in a few samples located near the plant, because alpha-spectrometry lacks in sensitivity and energy resolution to measure the $^{240}\text{Pu}/^{239}\text{Pu}$ isotopic ratio. As a beta-emitter, ^{241}Pu is not measurable by alpha-spectrometry and its analysis by beta-spectrometry has relatively poor detection limits (minimum of 10 mBq) ([Rosner et al., 1992](#)) compared to those obtained by mass spectrometry. ^{242}Pu has a very long half-life and cannot be measured efficiently by alpha-spectrometry at the ultra-trace level. Finally, ^{242}Pu is often used as an isotope dilution tracer for alpha spectrometry measurements.

In contrast, all plutonium isotopes, with the exception of ^{238}Pu (because of the presence of ^{238}U isotopes), can be measured by Inductively Coupled Plasma Mass Spectrometry (ICP-MS) with high accuracy. As mentioned in the literature review by Zheng *et al.* ([2013](#)), and summarized in Table S1 and Figure 1, ^{241}Pu and ^{242}Pu were only measured in a few studies, even though these isotopes are powerful tracers for identifying different sources of plutonium.

The objective of this current research was therefore to investigate the temporal variations of plutonium contamination in **flood** sediment in order to improve our understanding of plutonium dynamics in the FDNPP fallout-impacted region. As plutonium is strongly bound to the fine sediment mineral fraction ([Kersting, 2013](#); [Meusburger et al., 2016](#)), plutonium may be eroded, deposited and remobilised during rainfall events ([Ministry of Land Infrastructure and Transport, 2006](#); [Yoshimura et al., 2005](#)). In the Fukushima Prefecture, soil erosion and sediment transport were shown to be exacerbated during the typhoon season ([Chartin et al., 2013](#); [Evrard et al., 2013](#); [Yamashiki et al., 2014](#)).

Two types of samples were collected and analysed: i) **recently deposited flood sediment** collected at four sampling locations and in three different years to investigate the evolution of plutonium concentration and composition in a coastal catchment of the Fukushima Prefecture; and ii) soil samples collected at five sampling locations across the main radioactive plume to characterize the initial deposition of FDNPP-originating plutonium and compare its spatial pattern with that of the well-documented radiocaesium contamination plume ([Chartin et al., 2013](#)). To the best of our knowledge, this study represents the first attempt to characterise variations of plutonium contamination in sediment transiting a coastal river system draining the main radioactive plume in the Fukushima Prefecture.

Methodology

Study Area and Sampling

The main radioactive plume drained by Fukushima coastal rivers to the Pacific Ocean is located in the upper areas of these catchments (Figure 1). Contaminated sediment may therefore be eroded and transported to the coastal plains that were exposed to lower initial fallout levels. However, sediment was shown to be trapped in dams and reservoirs ([Lepage et al., 2016](#)), which delays their progressive transfer from contaminated upper catchment areas to the Pacific Ocean. Accordingly, this study will examine plutonium isotopic ratios in samples (including river sediment and soils) collected at different locations along the Niida River, where the only dam is located on a tributary and will therefore have a limited impact on sediment transport in this catchment ([Lepage et al., 2016](#)).

The Niida catchment (271 km²) is mainly covered with forests (75%), cultivated (12%) and built-up areas (22%) found in river valleys ([Evrard et al., 2016](#)). The soil sampling locations are located along the Niida River, in zones characterised by different levels of initial radiocaesium contamination (Figure 1).

The soil samples (top 1-2 cm) were collected in April 2012 and November 2012 (Figure 1, Table S2), following the protocol described elsewhere ([Lacey et al., 2016b](#)). In brief, soil was taken with a small plastic trowel and composed by 10 scrape subsamples. These subsamples were well mixed in the bag used to collect them. One sample was collected in the area characterised by the highest radiocaesium contamination (FSN 141). Three samples (FNS 116, FNS 135 and FNS 137) were taken in the moderate contaminated zone in the upper part of the catchment. The last sample (FNS 216) was collected in the less contaminated zone, in the downstream part of the catchment (Table S2).

The flood sediment sampling locations include two sites in the upper catchment ("Upstream Zone 1" and "Upstream Zone 2"), and two sites downstream ("Downstream Zone 1" and "Downstream Zone 2") (Figure 1). GPS coordinates of the sampling locations are provided in Table S2. River sediment samples were taken from material deposited after typhoons at these sites in November 2011, November 2013 and November 2014. These lag deposit samples correspond to fine particulate material that settled on channel banks, inset benches and floodplains. Care was taken to sample the fine material with a trowel or, when those deposits were too thin, with a spatula (down to the underlying coarser material layer). These recent flood deposits have proven to be comparable to in-situ suspended sediment samples in recent research conducted in similar erosive environments ([Lal et al., 2015](#); [Olley et al., 2013](#)). Each sample was composed of at least 5 subsamples collected over a 5 m reach using a plastic trowel. The

homogeneity of the radiocesium contamination across these river sections was verified in the field through the measurement of radiation rates using a radiameter on recent drape deposits in a 10 to 20 m wide area along rivers (Evrard et al., 2014a; Evrard et al., 2013). For the Downstream Zone 2 sampling point, located near the estuary of the Niida River, flood sediment was sampled on the alluvial plain terrace that is not subject to daily tides in order to limit the potential impact of seawater plutonium (Buet al., 2014). The potential heterogeneity of plutonium deposition was investigated by subdividing one sample (FNL 492) into three aliquots, which were measured for plutonium isotopic ratios and concentrations with the same protocol as the other samples.

Sample preparation

For each chemical preparation, 3 process blanks were prepared with the samples. Sample preparation (i.e. dissolution, separation and purification) is described in detail in a previous paper (Evrard et al., 2014b). In summary, samples were well mixed, dried in the oven at 40°C for 72 h and sieved to 2 mm. Then they were ground to a fine powder using an agar mortar and stored at room temperature before analysis. A ~5 g aliquot was transferred to a Pyrex beaker, covered with watch glasses and reduced to ash at 450°C for ~15 h in an electric furnace to decompose organic matter. After cooling to room temperature, in order to avoid interferences of isotopic impurities, a limited amount of ~100 fg of ^{244}Pu (diluted solution prepared from certified isotopic reference material IRMM-042a, Geel, Belgium) was added to the samples as an isotopic dilution tracer. Ashes were leached with concentrated HNO_3 (50 mL), evaporated to dryness, leached again with aqua regia (50 mL), evaporated to dryness, leached again with concentrated HCl (50 mL), evaporated to dryness, recovered with 2 M HCl and then filtered with a disposable 0.45 μm Nalgene filtering unit (Thermo Scientific, Rochester, NY, USA). The dissolved fraction was evaporated to dryness and recovered with concentrated HNO_3 (30 mL). A few mg of NaNO_2 were added to adjust and stabilize the redox state of plutonium at (+IV). After evaporation, 30 mL of 8 M HNO_3 were added to the sample prior to its loading onto a preconditioned ion-exchange resin column for plutonium purification.

Plutonium was first purified with a Dowex AG1X8 anion exchange resin filled with ~10 mL of 50-100 mesh AG1X8 resin then ~10 mL of 100-200 mesh AG1X8 resin, rinsed twice and conditioned with 8 M HNO_3 . After loading the sample solution onto the column, the resin was washed with 8 M HNO_3 (U-Am fraction elution), 10 M HCl (Th fraction elution) and NH_4I (1.5%) – 12 M HCl to elute plutonium from the resin. A second purification was performed with a Dowex AG1X4 anion-exchange resin (100-200 mesh resin), with the same wash operations and elution solution. Ion-exchange resins were used to purify and

concentrate plutonium from other potentially interfering elements (mainly U, Am and Pb). The final eluting solution was evaporated to dryness several times and recovered with concentrated HNO₃ to eliminate chloride. The last recovery was achieved with 3 mL of 2% HNO₃ for ICP-MS measurements. Overall, mean plutonium recovery was 72% (range: 39-83%, standard deviation: 14%). Plutonium purification used Dowex resins to completely remove ²⁴¹Am (>99%), which may interfere with the detection of ²⁴¹Pu by ICP-MS. As a trivalent ion, americium is not retained by the Dowex AG1X8 resin in nitric acid (Trémillon, 1965) (i.e. during the introduction of the sample into the column and the following washing step with 8M HNO₃).

Mass spectrometry for plutonium atom ratio and concentration measurements

Plutonium isotopic composition and concentrations were measured with a Multi-Collector ICP-MS (MC ICP-MS) ("Neptune Plus", Thermo-Fisher-Scientific., Bremen, Germany). The instrument was equipped with a Desolvating System (Cetac "Aridus II") and with high-efficiency cones ("Jet-cones", Thermo-Fischer-Scientific). Both equipments increased the sensitivity of the instrument by a factor of ~10. Five ion counters and eight Faraday cups are available for simultaneous measurement of all plutonium isotopes. Multi-collector measurements were performed in the static mode (Table S3).

Two certified reference materials were used: IRMM 186 (IRMM, Geel, Belgium) certified for ²³⁴U/²³⁸U, ²³⁵U/²³⁸U and ²³⁶U/²³⁸U isotopic ratios, and IRMM 057 certified for ²³³U concentration used as isotopic dilution tracer for uranium isotope ratio measurements. Mass bias, hydride formation rates, peak tails and gain factors of the ion counters were determined with a 5 ppb U solution of IRMM 186 and a calibrated mixed solution of IRMM 186 and IRMM 057. Both U standard solutions were diluted in 2% HNO₃. Results were systematically corrected for variation of the relative efficiency of the detectors (referred to as "gain factors" of the detectors) with respect to a selected Faraday cup. Relative standard deviations of the gain factors ranged between 0.3% and 5.1% during an analysis. These correction factors were applied to the measurement of a certified plutonium standard (CRM-128) during the analytical procedure (n>4), in order to verify the accuracy of the corrections (determined on a uranium standard) on plutonium samples.

Mass bias was corrected according to the ²³⁵U/²³⁸U value of the IRMM 186 standard measured before and after each sample analysis and applying an exponential law. Hydride formation rates and the peak tail at mass unit +1 were determined by measuring the (239 atomic mass unit)/²³⁸U ratio in the IRMM 186 standard solution (which does not contain plutonium), while the peak tail at mass unit +2 was

calculated by measuring the (240 atomic mass unit)/²³⁸U ratio in the same standard solution. Hydride and tail correction factors were systematically determined and applied. However, this correction did not exceed 0.4% in those results provided in this manuscript.

Detector configurations for measurements and gain factor calculations are shown in the Supplementary Material. ²⁴⁴Pu tracer isotopic impurities were corrected according to the certificate of the isotopic reference material IRMM-042a. Polyatomic interferences, mainly from PbO₂⁺ and PbAr⁺, were evaluated and corrected by measuring pure mono-elemental solutions of Pb at the end of the procedure. They were corrected through the calculation of their formation ratios and through the measurement of the count rates of Pb in the samples (Pointurier et al., 2011).

Finally, all plutonium isotopic ratios and plutonium concentrations provided in this study were corrected for the average plutonium content of the three chemical blanks. Isotopic ratios were then deduced from these concentrations and decay-corrected to March 15, 2011. Associated uncertainties were propagated during all calculations. Unless otherwise stated, all following uncertainties are expanded with a confidence level of 95% (e.g. a coverage factor of 2).

The validity of the correction method was verified by analysing several certified reference materials. IAEA-385 and IAEA-447 are certified for ²³⁹⁺²⁴⁰Pu mass activities. The results of the current research remained in good agreement with these certified values (Table S4). Ultra-traces of plutonium isotopes (mean total plutonium: 0.5 ± 3.4 fg) were detected in the process blanks, and these may originate from the different chemicals used for the additional leaching steps or subsequent contaminations during evaporation steps.

Protocol validation for plutonium extraction from sediment samples

In the protocol described above, it is assumed that plutonium is entirely extracted from the samples and insoluble fractions were therefore not analysed. In order to validate this assumption (i.e. the effectiveness of this protocol in recovering the plutonium from both global fallout and FDNPP sources from sediment samples) and consequently to evaluate the potential residual plutonium concentration in the insoluble fraction after filtration, an additional treatment was carried out on insoluble fractions to fully dissolve them. In brief, insoluble fractions and chemical blanks were transferred into clean and new Teflon beakers, spiked with ~70 fg of ²⁴²Pu (diluted solution prepared from a certified reference tracer, AEA Technology, UK), and submitted to 4 additional acid leaching steps: 3 with concentrated HF (40% weight) and concentrated HNO₃, and a 4th leaching with concentrated HNO₃ and H₂O₂. After each

leaching step, samples were evaporated to dryness on a hot plate (135°C covered with Teflon watch glass). At the end of this additional treatment, all samples were almost entirely dissolved, with the exception of a few refractory particles. A filtration was performed in order to remove the last visible solid residue. After carrying out similar plutonium purification with two Dowex anion exchange resins as for the soluble fraction, plutonium concentrations were measured in both soluble and insoluble fractions (Table S5).

Results showed that total plutonium masses in the soluble fractions ($^{239}\text{Pu} + ^{240}\text{Pu} + ^{241}\text{Pu}$) after acid leaching varied between 150 and 370 fg, while the total plutonium masses in the blanks were systematically lower than 2 fg for these measurements. In contrast, plutonium masses in the insoluble fractions ($^{239}\text{Pu} + ^{240}\text{Pu} + ^{241}\text{Pu}$) remained lower than 20 fg, representing less than 5% of the plutonium contained in the soluble fractions. These results confirm the effectiveness of the leaching dissolution protocol, which is sufficient to dissolve and recover a mean of $(96 \pm 1) \%$ of plutonium contained in the samples.

FDNPP and global fallout plutonium contributions

The fraction of plutonium in sediment (“sample”) originating from the FDNPP or the global fallout (“GF”) was calculated using a two-source mixing model (Kelley et al., 1999) with the respective isotopic abundances (Ab) of ^{241}Pu in the sample ($\text{Ab}(^{241}\text{Pu})_{\text{sample}}$), in the global fallout ($\text{Ab}(^{241}\text{Pu})_{\text{GF}}$) and at FDNPP ($\text{Ab}(^{241}\text{Pu})_{\text{FDNPP}}$). Isotopic abundances were calculated for each data set: the samples analysed in the current research, global fallout and FDNPP. Each data set corresponds to the isotopic ratios $^{240}\text{Pu}/^{239}\text{Pu}$, $^{241}\text{Pu}/^{239}\text{Pu}$ and $^{242}\text{Pu}/^{239}\text{Pu}$ measured in each sample in the current study (Table 2) or those determined for global fallout and FDNPP signatures and published in the literature (Table 1). $\text{Ab}(^{241}\text{Pu})_i$, where i represented either “sample” or “GF” or “FDNPP”, was calculated using Equation (1):

$$\text{Ab}(^{241}\text{Pu})_i = \frac{(^{241}\text{Pu}/^{239}\text{Pu})_i}{1 + (^{240}\text{Pu}/^{239}\text{Pu})_i + (^{241}\text{Pu}/^{239}\text{Pu})_i + (^{242}\text{Pu}/^{239}\text{Pu})_i} \quad (1)$$

where $(^{240}\text{Pu}/^{239}\text{Pu})_i$, $(^{241}\text{Pu}/^{239}\text{Pu})_i$ and $(^{242}\text{Pu}/^{239}\text{Pu})_i$ are atomic ratios.

Although plutonium isotopic ratios measured in soils before the FDNPP accident were shown to be rather homogeneous in the environment (Kelley et al., 1999), slight variations in the isotopic signature of plutonium emitted by the global fallout associated with nuclear tests were observed, and were shown to depend on the distance from the nuclear test sites, the proportion of stratospheric debris and their

location in the Southern or the Northern Hemisphere (Evrard et al., 2014b). Kelley et al. (1999), Zhang et al. (2010), Yang et al. (2015) and Muramatsu et al. (2003) reported isotopic ratios ($^{240}\text{Pu}/^{239}\text{Pu}$, $^{241}\text{Pu}/^{239}\text{Pu}$ and $^{242}\text{Pu}/^{239}\text{Pu}$) measured in soils collected worldwide before the FDNPP accident (Table 1).

Accordingly, the mean of these isotopic ratios was used to characterize the isotopic abundance associated with the global fallout: $\text{Ab}(^{241}\text{Pu})_{\text{GF}} = (1.27 \pm 0.13) \times 10^{-3}$.

Nishihara et al. (Nishihara et al., 2012), Kirchner et al. (2012) and Schwantes et al. (2012) estimated with the ORIGEN model the fuel compositions at the FDNPP at the moment of the accident (Table 1). The abundance of ^{241}Pu from the FDNPP was determined using the mean of these isotopic ratios: $\text{Ab}(^{241}\text{Pu})_{\text{FDNPP}} = (1.15 \pm 0.12) \times 10^{-1}$.

Finally, the fraction of plutonium originating from the FDNPP (X_{FDNPP}), i.e. the proportion of FDNPP-derived plutonium in each sample, was calculated for each sample using $\text{Ab}(^{241}\text{Pu})_{\text{sample}}$, $\text{Ab}(^{241}\text{Pu})_{\text{GF}}$ and $\text{Ab}(^{241}\text{Pu})_{\text{FDNPP}}$ in Equation (2).

$$\text{Ab}(^{241}\text{Pu})_{\text{FDNPP}} \times X_{\text{FDNPP}} + \text{Ab}(^{241}\text{Pu})_{\text{GF}} \times (1 - X_{\text{FDNPP}}) = \text{Ab}(^{241}\text{Pu})_{\text{sample}} \quad (2)$$

Radiocesium measurements and calculation of $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios

^{137}Cs activities were measured by gamma spectrometry using HPGe detectors (Canberra/Ortec) following the procedure described by Chartin et al. (2013). These ^{137}Cs measurements were used to calculate the $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios. In the literature, this ratio has been used to determine the amount of Pu released during the accident (Sakaguchi et al., 2014; Zheng et al., 2012b). In the current research, as the contributions of FDNPP-derived plutonium were determined for each sample (Equation 2), the corresponding activities of ^{239}Pu and ^{240}Pu originating from FDNPP were calculated for each individual sample and the $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios were determined with those FDNPP-derived plutonium activities.

Particulate matter properties and particle size parameters measurements

Particle size of the samples was analysed with laser granulometry (Malvern Mastersizer®) characterizing the textural parameters of particles ranging between 0.01 and 3500 μm , and the corresponding median diameters (d50) and Specific Surface Areas (SSA) of sediment samples were calculated. The EA-IRMS analyses were conducted with a continuous flow Elementar® VarioPyro cube analyzer coupled to a Micromass® Isoprime Isotope Ratio Mass Spectrometer (IRMS). The soil and sediment samples were

packed into tin capsules and all samples were first analyzed for Total Organic Carbon (TOC) and $\delta^{13}\text{C}$ using a set of tyrosine standards to calibrate TOC and Total Nitrogen (TN) concentrations and $\delta^{13}\text{C}$ measurements. C and $\delta^{13}\text{C}$ content were determined using an online continuous flow elemental analyzer Flash 2000 HT, coupled with an Isotopic Ratio Mass Spectrometer Delta V Advantage via a conflow IV interface from Thermo Fischer Scientific. C quantification was performed using a Certified reference material EMAP 2 purchased from IVA Analysentechnik (%C = 68.35 ± 0.28 %). Isotopic calibration was achieved using a Certified reference material: EMAP2 (-28.19 ± 0.14 ‰), and an Organic Analytical Standard (from Elemental Microanalysis): Sorghum Flour (-13.78 ± 0.17 ‰). In addition, a reference material LOS (Low Organic content Soil standard, from Elemental Microanalysis, %C= 1.61 ± 0.09 %, $\delta^{13}\text{C} = -26.66 \pm 0.24$ ‰) was used to assess the accuracy and reproducibility during the run. The standard deviation for LOS C content was of 1.65 ± 0.11 % (n=3). For $\delta^{13}\text{C}$ values, the standard deviation during the entire run was of 0.09 ‰ (n=8) for EMAP2, 0.29 ‰ (n=8) for Sorghum, and 0.21 ‰ (n=3) for LOS.

Results and Discussion

Pu isotopic measurements

Plutonium isotopic atom ratios and plutonium mass concentrations (fg of total plutonium per g of dry sediment) are provided in Table 2, while temporal variations of $^{240}\text{Pu}/^{239}\text{Pu}$, $^{241}\text{Pu}/^{239}\text{Pu}$ and $^{242}\text{Pu}/^{239}\text{Pu}$ in flood sediments are shown on Figure 2, Figure 3 and Figure 4, respectively. These plutonium isotopic ratios provided information regarding the origin and the fate of plutonium in the Niida River catchment. The proportions of plutonium originating from the FDNPP in each sample were then calculated from the isotopic abundances of ^{241}Pu (Figure 5 and Table 3).

Although at ultra-trace concentrations (<150 fg/g), the four plutonium isotopes (^{239}Pu , ^{240}Pu , ^{241}Pu and ^{242}Pu) were detected in all the samples, with one exception (soil sample FNS 216). The total plutonium masses measured for this sample remained in the same order of magnitude as those found in the three process blanks (4.5 ± 1.5 fg/g after blank correction). Accordingly, it was concluded that no plutonium was detected in this sample, which was therefore not considered when estimating the proportion of plutonium originating from the FDNPP.

In samples where plutonium was detected in significant concentrations, plutonium isotopic ratios generally plotted between the global fallout and FDNPP signatures (Figure 2, 3 and 4), with one exception for the $^{242}\text{Pu}/^{239}\text{Pu}$ atom ratio (FNL661 sample; Figure 4). These results demonstrate that most samples contained a mixture of plutonium from both sources. Following the FDNPP accident, plutonium deposition on soils was likely heterogeneous (Schneider et al., 2013). When examining plutonium in

three aliquots from one single sediment sample, plutonium concentrations remained similar. However, subsamples were characterized by differences in their isotopic ratios (Table 2). Although this sample was mixed and homogenised during the preparation process, plutonium from FDNPP remained heterogeneously distributed within the sample. This suggests that FDNPP-derived plutonium is contained in a limited number of plutonium-containing particles (Salbu, 2011) randomly distributed in each aliquot (Kimura et al., 2015; Yamamoto et al., 2014). This heterogeneity could explain the variations (~8%) observed in isotopic ratios when repeating the measurements on the same sample.

Spatial pattern of initial plutonium soil contamination

In all the soil samples (Figure 1, Table 2), the FDNPP-derived plutonium concentration increased with radiocaesium activities (correlation coefficient of 0.96, p -value = 0.03) (Figure S1). The highest FDNPP-originated plutonium concentration (27.5 ± 3.3 fg/g) was found in the sample FNS 141, which contained the highest ^{137}Cs activity (148.3 ± 1.5 kBq/kg). The second most contaminated sample in both plutonium and radiocaesium was sample FNS 135 (with a FDNPP-derived plutonium concentration of 2.3 ± 0.4 fg/g and a ^{137}Cs activity of 47.8 ± 0.5 kBq/kg). Sample FNS 116 contained 0.80 ± 0.33 fg/g of FDNPP-originated plutonium and 17.5 ± 0.2 kBq/kg of ^{137}Cs . Sample FNS 137 contained 0.34 ± 0.31 fg/g of FDNPP-derived plutonium and 8.8 ± 0.1 kBq/kg of ^{137}Cs . Finally, in the sample FNS 216 that was the less impacted by the radiocaesium deposition (1.2 ± 0.1 kBq/kg of ^{137}Cs), no plutonium was measured as already mentioned above, which suggests the absence of FDNPP-derived plutonium input at this location.

Temporal variations of plutonium in sediment

At the Upstream Zone 1 location, most isotopic ratios measured in sediment ($^{240}\text{Pu}/^{239}\text{Pu}$, $^{241}\text{Pu}/^{239}\text{Pu}$ and $^{242}\text{Pu}/^{239}\text{Pu}$) were higher than the global fallout signature (Table 1) (Muramatsu et al., 2003; Yang et al., 2015; Zhang et al., 2010), except for those $^{240}\text{Pu}/^{239}\text{Pu}$ ratios measured in sediment collected in November 2011 and November 2013. In November 2014, all isotopic ratios increased again at Upstream Zone 1 and they were even higher in 2014 than those found in 2011, suggesting a significant transfer of contaminated particles between November 2013 and November 2014 (Evrard et al., 2013; Laceby et al., 2016b). This site was the only investigated location potentially impacted by the remediation works (Evrard et al., 2016). In paddy fields, decontamination consisted of removing and replacing the uppermost 5 cm of topsoil with non-contaminated subsoil. Although decontamination should ultimately remove the most contaminated soil layer, there could be a significant plutonium remobilisation and transport phase from upper catchment areas when a typhoon occurs during the remediation works, as

vegetation must be cut in order to prepare these remediation works and soil is left bare and exposed to rainfall and runoff. It should be noted that the $^{242}\text{Pu}/^{239}\text{Pu}$ isotopic ratio found in 2014 was even higher than that associated with the FDNPP fallout ([Kirchner et al., 2012](#); [Nishihara et al., 2012](#); [Schwantes et al., 2012](#)), likely reflecting uncertainties on the estimation of the FDNPP signature or possible uncorrected interferences for the ^{242}Pu isotope (Table 1) ([Meusburger et al., 2016](#)).

At the second sampling location in the upper catchment area (Upstream Zone 2), plutonium masses remained very low (<21 fg/g) and stable with time. Moreover, all isotopic ratios did not change significantly for the three campaigns, with $^{241}\text{Pu}/^{239}\text{Pu}$ and $^{242}\text{Pu}/^{239}\text{Pu}$ remaining systematically higher than the global fallout signature. These results suggest that this area is less exposed to erosion, which is consistent with its location in the restricted access zone where soil cultivation was prohibited and where decontamination works started late in 2016. The material transiting in the river at this location, that may originate from nearby forests, was likely similar during the three campaigns.

At Downstream Zone 1, plutonium masses were very low (<20 fg/g) for the three measurements. However, all isotopic ratios decreased with time. The values found in 2011 were higher than the global fallout ratios, reflecting the fact that they contained a significant proportion of FDNPP-derived plutonium. Although the $^{241}\text{Pu}/^{239}\text{Pu}$ ratio in the sample collected in 2013 remained significantly higher than the global fallout signature, this was no longer the case in the sample collected in 2014. These results suggest a rapid and significant flush of contaminated material after typhoons in 2013 and 2014 ([Lacey et al., 2016a](#)). Overall, proportions of FDNPP-derived plutonium in these samples decreased by one order of magnitude with time, whereas the concentration of global fallout derived plutonium measured in 2014 was 45% higher than the one measured in 2011. The FDNPP-derived plutonium deposited on the uppermost soil surface layer (~1-2 cm) may have been washed away more rapidly compared to the global fallout plutonium that was likely homogenised in the soil profile (~15 cm) by the prolonged ploughing and puddling of the paddy fields during the last several decades ([Alewella et al., 2014](#); [Xu et al., 2016](#)).

Further downstream, at the Downstream Zone 2 sampling point, all isotopic ratios measured in 2013 were lower than those measured in 2011. However, no major difference was observed between the isotopic ratios measured in 2013 and 2014. Moreover, they were significantly higher than the global fallout signature, except for the $^{240}\text{Pu}/^{239}\text{Pu}$ ratio measured in samples collected in November 2013 and 2014. Total plutonium concentrations also decreased (-56%) between 2011 and 2014, further demonstrating the progressive flush of contaminated material. During the same period, an 85% decrease

in the fraction of FDNPP-derived plutonium was observed, whereas the fraction of global fallout plutonium only decreased by 43%, reflecting again the preferential export of FDNPP-derived plutonium immediately following the accident. This preferential export is likely due to the accumulation of FDNPP-derived plutonium in the upper soil layer, which is the most affected by erosion.

Comparison of radiocesium and plutonium activities

^{137}Cs levels measured after the FDNPP accident in soils and river sediment of the investigated region were up to 5 orders of magnitude higher ($10,000\text{--}100,000\text{ Bq kg}^{-1}$) than those prevailing before 2011, i.e. typically 30 Bq kg^{-1} (Fukuyama et al., 2010). Accordingly, all the ^{137}Cs measured in the samples analysed in the current research was considered to originate exclusively from the accident, which was confirmed by the measurements of ^{134}Cs and the calculation of $^{134}\text{Cs}/^{137}\text{Cs}$ activity ratios of $0.9\text{--}1.0$ (Chartin et al., 2013). Although radiocesium is rapidly bound to clay-sized sediment or material enriched in organic matter following the thermonuclear bomb testing (Motha et al., 2002; Tamura, 1964), these parameters were not found to provide the main factors controlling radionuclide levels measured in soils and sediment collected in Fukushima, as these were mainly driven by the spatial pattern of the initial radioactive fallout and its progressive erosion (Eyrolle-Boyer et al., 2016). Accordingly, in this post-accidental context, no significant relationship was found between ^{137}Cs concentrations and TOC (correlation coefficient of 0.43, $p\text{-value} = 0.02$) or SSA (correlation coefficient of 5×10^{-5}) in samples analysed in the current research. These findings are also consistent with previous results showing that radiocesium contamination was shown to be attached to multiple particle size fractions, including silt- and sand-sized particles (Sakaguchi et al., 2015).

$^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios in the soil samples (Table 2) remained in a very narrow range ($1.25 - 5.98 \times 10^{-7}$) suggesting that plutonium and radiocesium underwent similar deposition processes. In contrast, the corresponding values found in flood sediment deposits varied by several orders of magnitude (from 2.39×10^{-8} to 1.19×10^{-5}) reflecting the changes of source contributions to sediment transiting the river throughout time. Other $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios measured in environmental samples collected in the Fukushima region after the FDNPP accident were provided in the literature (Sakaguchi et al., 2014; Zheng et al., 2012b). Zheng et al. (2012b) compared this ratio with that measured after the Chernobyl accident. Yamamoto et al. (2014) estimated that this ratio varied between 2.6×10^{-9} and 4.8×10^{-6} in road dust, litter and soil samples. They observed a positive relationship between this ratio and the distance from the power plant. In the current study, no such relationship was observed, which is likely explained by the fact that all samples were collected relatively far from FDNPP (between 25 - 45 km).

However, the spatial variation of this ratio across the catchment may also provide additional information on the processes affecting sediment-bound radionuclide redistribution after the FDNPP accident. The increase in $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios observed at the Upstream Zone 1 sampling location (Table 2) suggests that the remobilisation of plutonium after the completion of remediation works differed from that of radiocesium. In contrast, the $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios at the upstream Zone 2 site did almost not change during the studied period (Table 2), confirming the previous conclusion that this sampling location is less exposed to erosion as it is located in a restricted zone and it is mainly covered with forests which are less sensitive to soil erosion. The same observation can be made for the Downstream Zone 1 location, with the absence of significant change ($p\text{-value}=0.06$) in the $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios found at this location between 2011 and 2014 (Table 2). This result suggested a simultaneous decrease of plutonium and radiocesium, in the same relative proportions. Finally, at the last sampling location (Downstream 2), the $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios also did not vary significantly ($p\text{-value}=0.08$), suggesting again a simultaneous decrease of both radionuclide contamination at this site, in the same proportions.

Conclusions and perspectives

In the current research, plutonium concentrations and isotopic ratios were measured in soil and flood sediment collected at various locations in the Niida catchment after the FDNPP accident. Plutonium isotopic ratios associated with the global fallout were shown to be homogeneous and these ratios were therefore used to calculate the respective sources of Pu, in particular the contribution of FDNPP to the total amount of Pu found in these sediment samples. As demonstrated by the soil analyses, the spatial pattern of initial plutonium deposition was similar to that of radiocaesium (Figure 1). In November 2011, isotopic ratios and FDNPP derived-plutonium concentrations were higher in the coastal plain sediment samples than those obtained in upper catchment areas. This suggests that FDNPP plutonium, eroded from the topsoil and transported with sediment particles during floods, was rapidly transported downstream. The analysis of the $^{239+240}\text{Pu}/^{137}\text{Cs}$ activity ratios found in the soil samples confirmed that radiocesium and plutonium underwent similar deposition processes. In general, their redistribution with sediment in the catchment followed the same trends, except at the site affected by remediation works.

Although the sampling procedure was rigorously the same during each field campaign and the samples were well-mixed, future research should examine the variability of these sediment deposits along a river reach. Moreover, the number of investigated sites remained limited. The main challenge when analysing the spatial-temporal variations of actinides is related to the very long time (e.g. several weeks) needed to

process samples and analyse them for plutonium with MC-ICP-MS. The ongoing development of more rapid measurement techniques may provide solutions in the future for analysing larger sets of samples. For example, some authors recently proposed several improvements in dissolution methods and in chemical separations, which are faster and more efficient ([Bu et al., 2017](#); [Men et al., 2018](#)).

Different plutonium isotopic ratios measured in three homogenised aliquots from one sample, suggested the emission of discrete plutonium-bearing particles during the FDNPP accident (Table 4). This result also shows the importance of processing representative samples with a significant quantity of material (e.g. 5 g samples in the current research, while others studies compiled in Table S1 processed 1 to 5 g of samples) as plutonium contamination from FDNPP may be in the form of a limited number of discrete plutonium-bearing particles. The characterisation of these particles requires further investigation. The particulate form of FDNPP-derived plutonium must be examined in order to better understand the origin of plutonium from the power plant (mixed-oxide fuel or irradiated UO₂ fuel) and to improve our knowledge of the mechanisms releasing actinides during the FDNPP accident (shape and composition of the emitted particles). These properties should also impact the fate and the residence time of actinides in the environment.

Accordingly, to further understand the migration of plutonium in the fallout-impacted region, screening methods should be developed and applied to identify, isolate and characterize plutonium-bearing particles. Elemental and isotopic compositions of these particles should then be measured to determine their source (e.g. formation mechanism in the reactor, source reactor, etc.) and to characterise their fate in the environment (e.g. ability to dissolve, to be remobilized, to be associated with colloids, etc.).

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Tables

Table 1: Estimates of the plutonium isotopic atom ratios in FDNPP at the time of the reactor shutdown and measured isotopic ratios of fallout plutonium in soils of the region representative of global fallout. Uncertainties associated with ratios of global fallout are extended with a confidence level of 95% (coverage factor of 2).

		$^{240}\text{Pu}/^{239}\text{Pu}$	$^{241}\text{Pu}/^{239}\text{Pu}$	$^{242}\text{Pu}/^{239}\text{Pu}$
FDNPP	Nishihara et al. (2012)			
	Reactor 1	0.344	0.1921	0.0662
	Reactor 2	0.320	0.1922	0.0630
	Reactor 3	0.356	0.1827	0.0620
	Schwantes et al. (2012)	0.447	0.1962	
	Kirchner et al. (2012)	0.395	0.1738	
Mean		0.373 ± 0.044	0.1874 ± 0.0081	0.0637 ± 0.0026
Global Fallout	Kelley et al. (1999)			
	Av. soils NH 30°-71°	0.180 ± 0.014	0.0011 ± 0.0001	0.0039 ± 0.0007
	Sapporo, Japan	0.177 ± 0.002	0.0011 ± 0.0001	0.0038 ± 0.0002
	Tokyo, Japan	0.176 ± 0.002	0.0010 ± 0.0001	0.0035 ± 0.0002
	Zhang et al. (2010)	0.192 ± 0.003	0.0026 ± 0.0003	
	Zhang et al. (2010)	0.192 ± 0.004	0.0029 ± 0.0006	
	Muramatsu et al. (2003)	0.179 ± 0.001		
	Yang et al. (2015)	0.183 ± 0.011		
Mean		0.181 ± 0.005	0.0015 ± 0.0003	0.0037 ± 0.0005

485 Table 2: Plutonium concentrations and atom ratios measured in soil and sediment samples collected in the Niida River. Extended uncertainties with a
 486 confidence level of 95% (coverage factor of 2); $^{241}\text{Pu}/^{239}\text{Pu}$ values were decay-corrected to 15 March 2011. ^{240}Pu , ^{241}Pu and ^{242}Pu isotopes were not
 487 detected in samples FNS216 and FNL477.

Sample			Location	$^{240}\text{Pu}/^{239}\text{Pu}$ atom ratio	$^{241}\text{Pu}/^{239}\text{Pu}$ atom ratio	$^{242}\text{Pu}/^{239}\text{Pu}$ atom ratio	Total plutonium mass concentration (fg/g)	$^{239+240}\text{Pu}/^{137}\text{Cs}$ (activity ratio)
Soil	FNS116	Apr.2012	-	0.188 ± 0.007	0.0041 ± 0.0009	0.006 ± 0.002	45.7 ± 1.3	$1.48 \times 10^{-7} \pm 4.33 \times 10^{-8}$
	FNS135	Apr.2012		0.190 ± 0.002	0.0069 ± 0.0004	0.007 ± 0.003	62.9 ± 7.2	$1.58 \times 10^{-7} \pm 1.99 \times 10^{-8}$
	FNS137	Apr.2012		0.183 ± 0.012	0.0033 ± 0.0014	0.005 ± 0.001	29.7 ± 0.6	$1.25 \times 10^{-7} \pm 8.25 \times 10^{-8}$
	FNS141	Apr.2012		0.205 ± 0.008	0.0287 ± 0.0018	0.013 ± 0.001	145.2 ± 1.5	$5.98 \times 10^{-7} \pm 5.17 \times 10^{-8}$
	FNS216	Nov.2012		-	-	-	4.6 ± 1.5	-
Flood Sediment	FNL058 A	Nov.2011	Upstream Zone 1	0.188 ± 0.016	0.0074 ± 0.0042	0.012 ± 0.006	54.5 ± 3.4	$1.01 \times 10^{-7} \pm 5.46 \times 10^{-8}$
	FNL477	Nov.2013		0.174 ± 0.012	0.0186 ± 0.0036	0.0106 ± 0.0038	44.4 ± 2.31	$2.19 \times 10^{-6} - 5.15 \times 10^{-7}$
	FNL661	Nov.2014		0.267 ± 0.016	0.0668 ± 0.0100	0.116 ± 0.014	44.5 ± 2.9	$1.19 \times 10^{-5} \pm 1.61 \times 10^{-6}$
	FNL031 A	Nov.2011	Upstream Zone 2	0.195 ± 0.028	0.0075 ± 0.0019	0.016 ± 0.008	14.6 ± 0.6	$2.39 \times 10^{-8} \pm 6.57 \times 10^{-9}$
	FNL492	Nov.2013		0.186 ± 0.027	0.0066 ± 0.0015	0.019 ± 0.007	16.2 ± 0.7	$3.33 \times 10^{-8} \pm 8.04 \times 10^{-9}$
	FNL667	Nov.2014		0.183 ± 0.026	0.0069 ± 0.0023	0.018 ± 0.010	20.3 ± 0.9	$4.23 \times 10^{-8} \pm 1.41 \times 10^{-8}$
	FNL064	Nov.2011	Downstream Zone 1	0.217 ± 0.023	0.0447 ± 0.0127	0.037 ± 0.012	17.1 ± 2.6	$3.33 \times 10^{-7} \pm 8.07 \times 10^{-8}$
	FNL468	Nov.2013		0.193 ± 0.028	0.0135 ± 0.0078	0.008 ± 0.005	16.0 ± 0.7	$1.01 \times 10^{-7} \pm 4.79 \times 10^{-8}$
	FNL685	Nov.2014		0.173 ± 0.022	0.0032 ± 0.0019	0.004 ± 0.002	17.8 ± 0.8	$1.09 \times 10^{-7} \pm 1.05 \times 10^{-7}$
	FNL043 B	Nov.2011	Downstream Zone 2	0.223 ± 0.007	0.0464 ± 0.0024	0.022 ± 0.002	94.8 ± 15.9	$6.52 \times 10^{-6} \pm 9.40 \times 10^{-7}$
	FNL465	Nov.2013		0.183 ± 0.007	0.0097 ± 0.0018	0.010 ± 0.002	48.1 ± 8.6	$4.19 \times 10^{-7} \pm 9.16 \times 10^{-8}$
	FNL688	Nov.2014		0.174 ± 0.008	0.0158 ± 0.0041	0.009 ± 0.003	42.0 ± 7.7	$6.23 \times 10^{-6} \pm 1.60 \times 10^{-6}$

489 Table 3: Relative contributions and concentrations of FDNPP and global fallout plutonium in the analysed
 490 samples, modelled with ^{241}Pu measurements. Uncertainties are extended with a confidence level of 95%
 491 (coverage factor of 2).

Sample		Date of sampling	Location	Proportion of Plutonium from FDNPP			Plutonium concentration from FDNPP in the samples (fg/g)			Plutonium concentration from global fallout in the samples (fg/g)		
Soil	FNS116	Apr.2012	-	1.9 %	±	0.7 %	0.9	±	0.3	44.8	±	1.3
	FNS135	Apr.2012		3.9 %	±	0.5 %	2.4	±	0.4	60.5	±	7.3
	FNS137	Apr.2012		1.3 %	±	1.1 %	0.4	±	0.3	29.3	±	0.6
	FNS141	Apr.2012		19.1 %	±	2.3 %	27.7	±	3.3	117.5	±	3.7
	FNS216	Nov.2012		-			-			-		
Flood Sediment	FNL058A	Nov. 2011	Upstream Zone 1	4.3 %	±	3.1 %	2.3	±	1.7	52.1	±	3.8
	FNL477	Nov. 2013		12.4 %	±	2.9 %	5.5	±	1.3	38.9	±	2.7
	FNL661	Nov. 2014		39.3 %	±	7.0 %	17.5	±	3.3	27.0	±	4.4
	FNL031A	Nov. 2011	Upstream Zone 2	3.7 %	±	1.4 %	0.5	±	0.2	14.0	±	0.7
	FNL492	Nov. 2013		3.6 %	±	1.2 %	0.6	±	0.2	15.6	±	0.7
	FNL667	Nov. 2014		3.9 %	±	1.7 %	0.8	±	0.4	19.5	±	0.9
	FNL064	Nov. 2011	Downstream Zone 1	29.0 %	±	8.8 %	5.0	±	1.7	12.1	±	3.1
	FNL468	Nov. 2013		8.6 %	±	5.6 %	1.4	±	0.9	14.6	±	1.2
	FNL685	Nov. 2014		1.2 %	±	1.4 %	0.2	±	0.3	17.6	±	0.8
	FNL043B	Nov. 2011	Downstream Zone 2	30.3 %	±	3.4 %	28.8	±	5.8	65.9	±	17.0
	FNL465	Nov. 2013		5.9 %	±	1.4 %	2.9	±	0.9	45.2	±	8.6
	FNL688	Nov. 2014		10.4 %	±	3.1 %	4.4	±	1.5	37.6	±	7.8

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493 Table 4: Plutonium isotopic ratios and concentrations for a sample divided into three aliquots.
 494 Uncertainties are extended with a confidence level of 95% (coverage factor of 2)

Sample	Date of sampling	Location	$^{240}\text{Pu} / ^{239}\text{Pu}$ atom ratio	$^{241}\text{Pu} / ^{239}\text{Pu}$ atom ratio	$^{242}\text{Pu} / ^{239}\text{Pu}$ atom ratio	Total plutonium mass concentration (fg/g)
FNL492 - 1	Nov.2013	Upstream Zone 2	0.188 ± 0.027	0.0145 ± 0.0029	0.048 ± 0.012	14.82 ± 0.62
FNL492 - 2	Nov.2013		0.184 ± 0.026	0.0069 ± 0.0018	0.018 ± 0.008	15.81 ± 0.68
FNL492 - 3	Nov.2013		0.177 ± 0.026	0.0081 ± 0.0022	0.010 ± 0.009	17.92 ± 0.78

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499   Figure 1: Location of soil and sediment samples containing FDNPP-derived plutonium from literature and
500   location of samples analysed in this current research in the Niida catchment delineated in black. The
501   main radiocaesium contamination plume is also located in the background map derived from Chartin et
502   al. (2013).

503

Figure 2: $^{240}\text{Pu}/^{239}\text{Pu}$ atomic ratios measured in this study. The blue line corresponds to the mean global fallout value (Kelley et al., 1999; Muramatsu et al., 2003; Yang et al., 2015; Zhang et al., 2010), the red line corresponds to estimations of the FDNPP signature (Kirchner et al., 2012; Nishihara et al., 2012; Schwantes et al., 2012). Uncertainties are extended with a confidence level of 95% (coverage factor of 2).

Figure 3: $^{241}\text{Pu}/^{239}\text{Pu}$ atomic ratios measured in this study. The blue line corresponds to the mean global fallout value (Kelley et al., 1999; Zhang et al., 2010), the red line corresponds to estimations of the FDNPP signature (Kirchner et al., 2012; Nishihara et al., 2012; Schwantes et al., 2012). Uncertainties are extended with a confidence level of 95% (coverage factor of 2). Numbers of ^{241}Pu atoms were decay-corrected to 15 March 2011.

Figure 4: $^{242}\text{Pu}/^{239}\text{Pu}$ atomic ratios measured in this study. The blue line corresponds to the mean global fallout value (Kelley et al., 1999), the red line corresponds to estimations of the FDNPP signature (Kirchner et al., 2012; Nishihara et al., 2012; Schwantes et al., 2012). Uncertainties are extended with a confidence level of 95% (coverage factor of 2).

Figure 5: Total plutonium concentrations in all samples (red dots) and contributions of FDNPP-originating plutonium (blue histograms) deduced from the ^{241}Pu abundances in each sample. Uncertainties are extended with a confidence level of 95% (coverage factor of 2).

527 **Supplementary Information:**

528 Table S1: Plutonium isotopic ratios measured in environmental samples collected in Japan after the FDNPP accident (PPG refers to the Pacific
529 Proving Ground).

Measurement technique	Source	Sample types	Sampling date	Direction (and distance from FDNPP in km)	²⁴⁰ Pu/ ²³⁹ Pu atom ratio	²⁴¹ Pu/ ²³⁹ Pu atom ratio	²⁴¹ Pu/ ²³⁹ + ²⁴⁰ Pu activity ratio	²³⁸ Pu/ ²³⁹ + ²⁴⁰ Pu activity ratio
ICP-MS	Zheng <i>et al.</i> (2012b)	Litter and surface soil (n=7)	April – Aug., 2011	NW (26–32)	0.323-0.330	0.128-0.135	107.8	-
		Soil (n=12)		S (20)	0.303	0.103	-	-
				SW (130–230)	0.171 - 0.303	-	-	-
	Schneider <i>et al.</i> (2013)	Soil (n=13) and plant (n=7)	Oct. – Dec 2011	NW and SW (1 – 244)	0.381 - 0.64	-	-	-
	Evrard <i>et al.</i> (2014b)	River sediment (n=6)	Nov. 2011 – May 2013	NW (10–70)	0.150 - 0.281	0.0019-0.100	2.0–82.3	-
	Zheng <i>et al.</i> (2012a)	Marine surface sediments (n=9)	July – Aug. 2011	NE (200)	0.188, 0.189	0.0034	-	-
				E (30–200)	0.22 - 0.26 (PPG)	-	-	-
	Sakaguchi <i>et al.</i> (2012)	River water (n=1)	June – Aug. 2011	NW (70)	0.308	-	-	-
		Seawater (n=2)		E (30)	0.181 - 0.218 (PPG)	-	-	-
Alpha-spectrometry	MEXT(2011)	Surface soils (n=5)	June – July 2011	NW (5–100)	-	-	-	0.33 – 2.2
	Yamamoto <i>et al.</i> (2014)	Soil (n=8)	2011 – 2012	NW (20–30)	-	-	-	0.088-1.324
	Imanaka <i>et al.</i> (2012)	Soil (n=3)	28 – 29 March 2011	NW (30)	-	-	-	0.110-1.324
	Utsunomiya <i>et al.</i> (2012)	Sediment (n=5)	No information	NW (3)	-	-	-	0.319-2.60
	Kimura <i>et al.</i> (2015)	Soil (n=6)	No information	NW (4)	-	-	-	0.22-1.54

531 Table S2: Location of the investigated samples.

Sample		Date of sampling	Location	GPS coordinates	
				Y	X
Soil	FNS116	Apr.2012	-	37.693042	140.698035
	FNS135	Apr.2012		37.662240	140.728147
	FNS137	Apr.2012		37.662406	140.791832
	FNS141	Apr.2012		37.624151	140.810154
	FNS216	Nov.2012		37.63945	140.933293
Flood Sediment	FNL058A	Nov.2011	Upstream Zone 1	37.690509	140.676251
	FNL477	Nov.2013			
	FNL661	Nov.2014			
	FNL031A	Nov.2011	Upstream Zone 2	37.61385	140.800832
	FNL492	Nov.2013			
	FNL667	Nov.2014			
	FNL064	Nov.2011	Downstream Zone 1	37.651338	140.958329
	FNL468	Nov.2013			
	FNL685	Nov.2014			
	FNL043B	Nov.2011	Downstream Zone 2	37.643776	141.00301
	FNL465	Nov.2013			
	FNL688	Nov.2014			

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Detector configurations

Gain corrections were determined with minor U isotopes in calibration standard. Accordingly, measurements were performed in the same level of ion beams signals as those of Pu isotopes in the samples. This avoided the potential problems of non-linearity of the ion counters. Moreover, for determining gain factors, the U concentration level was chosen in order to measure the $^{235}\text{U}/^{238}\text{U}$ ratio in Faraday cups. An “internal” mass bias factor could then be determined. This internal mass bias allowed to correct for mass bias without using those corrections determined with standard bracketing. The ion counter IC1, IC3 and IC4 gain factors were respectively determined with $^{236}\text{U}/^{238}\text{U}$, $^{234}\text{U}/^{238}\text{U}$ and $^{233}\text{U}/^{238}\text{U}$ ratios in a mixed solution of uranium standards IRMM186 and IRMM057. The IC2 gain factor was determined with ^{235}U measured once in IC2 and once in L5 cup. Finally, the IC5 gain factor was calculated with a third measurement of $^{235}\text{U}/^{238}\text{U}$ conducted on the mixed solution of IRMM 186 and IRMM 057.

Table S3: Detector configurations for plutonium samples and uranium standard measurements on MC ICP-MS. Isotopes measured on ion counting detectors are presented in bold. a: for measurement of peak tail at m+1 + uranium hydride ratio, b: for measurement of peak tail at m+2.

Cup		IC4	IC3	IC2/L5	IC1	L4	IC5	L2	H1	H3
Pu samples		^{239}Pu	^{240}Pu	^{241}Pu	^{242}Pu		^{244}Pu			
IRMM 186	Mass bias								^{235}U	^{238}U
	Hydrides, tail	$^{239}\text{Pu}^a$	$^{240}\text{Pu}^b$							
IRMM 186 + IRMM 057	Gain Factor IC1, IC3, IC4	^{233}U	^{234}U	^{235}U	^{236}U	^{238}U				
	Gain Factor IC2	^{233}U	^{234}U	^{235}U	^{236}U	^{238}U				
	Gain Factor IC5						^{235}U	^{238}U		

Table S4: Comparison of plutonium measurements and certified values in IAEA reference materials. Uncertainties are extended with a confidence level of 95% (coverage factor of 2).

Material	Measured $^{239+240}\text{Pu}$	Certified $^{239+240}\text{Pu}$
IAEA – 385	2.83 ± 0.19	2.89 – 3.00
IAEA – 447	3.70 ± 1.77	4.90 – 5.70

Table S5: Calculated plutonium masses in process blanks and in soluble and insoluble fractions after blank corrections. Uncertainties are extended with a confidence level of 95% (coverage factor of 2).

	Sample masses (g)	Plutonium masses in the soluble fraction (fg)			Plutonium masses in the insoluble fraction (fg)		
Process Blank 1	-	1.2	±	0.4	0.3	±	0.3
Process Blank 2	-	0.3	±	1.2	0.6	±	0.3
Process Blank 3	-	1.9	±	0.5	0.5	±	0.3
Average blank		1.1	±	0.4	0.5	±	0.4
FNL 043B	3.5	324	±	56	14.9	±	2.3
FNL 465	3.9	184	±	33	7.0	±	1.6
FNL 688	4.7	196	±	36	4.9	±	1.0

Table S6 : Activities in $^{239+240}\text{Pu}$ and ^{137}Cs in soil and sediment samples collected in the Niida River catchment. Extended uncertainties with a confidence level of 95% (coverage factor of 2).

Sample		Date of sampling	Location	$^{239+240}\text{Pu}$ (Bq/kg)			^{137}Cs (kBq/kg)		
Soil	FNS116	Apr.2012	-	0.003	±	0.001	17.5	±	0.2
	FNS135	Apr.012		0.008	±	0.001	47.8	±	0.5
	FNS137	Apr.2012		0.001	±	0.001	8.8	±	0.1
	FNS141	Apr.2012		0.47	±	0.008	148.3	±	1.5
	FNS216	Nov.2012		-			1.2	±	0.1
Flood Sediment	FNL058A	Nov.2011	Upstream Zone 1	0.007	±	0.004	71.0	±	0.5
	FNL477	Nov.2013		-			8.0	-	0.2
	FNL661	Nov.2014		0.055	±	0.007	4.6	±	0.1
	FNL031A	Nov.2011	Upstream Zone 2	0.002	±	0.000	73.6	±	0.5
	FNL492	Nov.2013		0.002	±	0.000	54.0	±	0.5
	FNL667	Nov.2014		0.002	±	0.001	57.2	±	0.5
	FNL064	Nov.2011	Downstream Zone 1	0.016	±	0.004	47.1	±	0.3
	FNL468	Nov.2013		0.004	±	0.002	43.5	±	0.4
	FNL685	Nov.2014		0.001	±	0.001	5.6	±	0.1
	FNL043B	Nov.2011	Downstream Zone 2	0.093	±	0.013	14.2	±	0.3
	FNL465	Nov.2013		0.009	±	0.002	21.1	±	0.2
	FNL688	Nov.2014		0.014	±	0.003	2.2	±	0.1

Table S7 : Particulate matter properties and particle size parameters measured in soil and river sediment samples collected in the Niida River catchment. D50 is the median diameter of the grains. SSA is the Specific Surface Area. Extended uncertainties with a confidence level of 95% (coverage factor of 2). n.a.: not available.

	Sample	Date of sampling	Location	Total Organic Carbon (%)	$\delta^{13}\text{C}$ (‰)	D50 (μm)	SSA (m^2/kg)
Soil	FNS116	Apr.2012	-	1.97 $\pm$ 0.05	-25.84	79.90	207.93
	FNS135	Apr.012		6.11 $\pm$ 0.11	-25.73	189.33	136.33
	FNS137	Apr.2012		3.29 $\pm$ 0.08	-26.23	96.70	207.93
	FNS141	Apr.2012		7.75 $\pm$ 0.14	-27.77	64.00	349.47
	FNS216	Nov.2012		2.11 $\pm$ 0.06	-26.47	137.50	214.35
Flood Sediment	FNL058A	Nov.2011	Upstream Zone 1	n.a.	n.a.	n.a.	n.a.
	FNL477	Nov.2013		5.00 $\pm$ 0.10	-27.10	46.23	325.1
	FNL661	Nov.2014		3.96 $\pm$ 0.10	-27.09	75.13	198.97
	FNL031A	Nov.2011	Upstream Zone 2	3.47 $\pm$ 0.08	-26.83	235	48.91
	FNL492	Nov.2013		4.04 $\pm$ 0.09	-27.51	175.00	80.72
	FNL667	Nov.2014		4.80 $\pm$ 0.09	-27.01	118.00	92.29
	FNL064	Nov.2011	Downstream Zone 1	4.57 $\pm$ 0.11	-28.37	n.a.	n.a.
	FNL468	Nov.2013		2.67 $\pm$ 0.10	-28.20	208.67	93.44
	FNL685	Nov.2014		8.47 $\pm$ 0.07	-27.31	n.a.	n.a.
	FNL043B	Nov.2011	Downstream Zone 2	6.73 $\pm$ 0.14	-27.80	n.a.	n.a.
	FNL465	Nov.2013		2.38 $\pm$ 0.12	-28.06	n.a.	n.a.
	FNL688	Nov.2014		3.47 $\pm$ 0.06	-24.37	130.00	238.53

Figure S1: Correlation between FDNPP-derived plutonium concentration and ^{137}Cs activities in soil samples. Uncertainties are extended with a confidence level of 95% (coverage factor of 2).

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