

Introducing Tritium (^3H) MAS NMR in biosolids: application to crystalline diphenylalanine

Thibault Charpentier, Céline Taglang, Marielle Kenfack, Gregory Pieters,
Sophie Feuillastre, Yves Boulard, Marie-Thérèse Paternostre, Christophe
Dugave, Bernard Rousseau

► **To cite this version:**

Thibault Charpentier, Céline Taglang, Marielle Kenfack, Gregory Pieters, Sophie Feuillastre, et al..
Introducing Tritium (^3H) MAS NMR in biosolids: application to crystalline diphenylalanine. EURO-
MAR 2018, Jul 2018, Nantes, France. cea-02340731

HAL Id: cea-02340731

<https://hal-cea.archives-ouvertes.fr/cea-02340731>

Submitted on 31 Oct 2019

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Solid-state NMR methods

EUROMAR2018/407

Introducing Tritium (^3H) MAS NMR in biosolids: application to crystalline diphenylalanine

Thibault Charpentier¹, Céline Taglang², Marielle Tamigney Kenfack², Gregory Pieters², sophie feuillastre², Yves Boulard³, Marie-Thérèse Paternostre³, Christophe Dugave², Bernard Rousseau²

¹NIMBE, ²SCBM, ³IB2C, CEA SACLAY, Gif-sur-Yvette, France

If selected, I agree to present my work during an oral communication: Yes

Abstract: Tritium (^3H) possesses the highest gyromagnetic ratio of the periodic table and its negligible natural abundance ($3 \times 10^{-16} \%$) ensures that no background signal is observed, in contrast to ubiquitous proton (^1H), so that one can work at a very low concentration (in the present work less than 0.1%) while maintaining a very good sensitivity. Despite these appealing features, only very few ^3H MAS NMR study has been reported so far [1], probably due to the safety issues that need to be carefully addressed for rotating a radioactive sample. We have recently developed an appropriate NMR instrumentation (using a DOTY $^3\text{H}/^1\text{H}/\text{X}$ XC4 probehead) and demonstrated the feasibility of ^3H MAS NMR experiments on model compounds: distance measurements up to 14 Å could be reached and medium-range distances (4-8 Å) could be determined with an unprecedented resolution (0.02 Å) [2]. In this work, we extend these experiments to a relevant system for biological studies, the diphenylalanine peptide which has recently attracted great attention for the development of DNP-based ^{13}C - ^{13}C distances measurements at natural abundance [3]. Reliable distances could be determined up to 7 Å so that our methodology can perfectly complement these experiments.

Regarding NMR methodologies, ^1H - ^3H Cross Polarization was found to work very effectively and yielded a significant enhancement of the signal, mainly as a consequence of the much shorter relaxation time of protons. A moderate spinning rate of typically 8 kHz was found to be sufficient in obtaining narrow ^3H MAS NMR peaks with an excellent resolution of 0.01-0.02 ppm. Specificities of the strongly coupled ^1H - ^3H nuclear spins system will be discussed, especially with regards to the choice of the pulse sequences for ^3H - ^3H dipolar recoupling where HORROR was found to be the most efficient one. Indeed, it requires only a moderate ^3H RF field while strong ^1H decoupling can be applied, thus keeping the total RF power reasonable on the two channels which are very close in frequency. Our distance measurements were combined to Molecular Dynamics simulations and DFT computations of NMR shifts in the perspective to develop an integrated ^3H MAS NMR methodology to determine conformation of a small molecule self-assembled to form large-architecture.

[1]. J.P. Bloxsidge et al. *Solid State NMR of Polymers*, I. Ando, T. Asakura Editors, Studies in Physical and Theoretical Chemistry, Vol. 84, Chapter 6.3, 1998 Elsevier science. B.V.

[2] A.K.L. Yuen et al., *J. Am. Chem. Soc.* 132, 1734-1735 (2010).

[3] K. Märker et al., *Chemical Science* 8 974-987 (2016).

You can insert an image with the button on the right:

