Efficient reductive depolymerization of hardwood and softwood lignins with Brookhart’s iridium(III) catalyst and hydrosilanes

Louis Monsigny, Elias Feghali, Jean Claude Berthet, Thibault Cantat

To cite this version:


HAL Id: cea-01760087
https://hal-cea.archives-ouvertes.fr/cea-01760087
Submitted on 6 Apr 2018

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L’archive ouverte pluridisciplinaire HAL, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d’enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.
Efficient reductive depolymerization of hardwood and softwood lignins with Brookhart’s iridium(III) catalyst and hydrosilanes

Louis Monsigny,a Elias Feghali,a Jean-Claude Berthet,a and Thibault Cantat*a

Efficient catalytic reduction of lignin model molecules and reductive depolymerization of softwood and hardwood lignins is presented with the iridium based Brookhart’s catalyst and hydrosilanes RSiH as reductant. This catalyst displays increased stability and selectivity in comparison to the B(C6F5)3/hydrosilane system and it enables a convergent reductive depolymerization of wood lignins to isolable mono-aromatics.

Introduction

Accounting for 15-30 % of organic carbon in the biosphere, lignin is a prominent biopolymer, present in wood, plants, and agricultural residues. Because it derives from aromatic monolignols (Figure 1), this material can be viewed as a large and sustainable source of aromatic chemicals that could advantageously replace petrochemicals in the long run.1-2 Whereas lignin is currently burnt to supply energy in the pulp and paper industry, the chemical valorization of this biopolymer requires the selective deconstruction of its polymeric structure into a narrow range of molecules. Such endeavor remains a challenge, but the chemical stability and heterogeneity of lignin and the lack of robust and active catalysts as well as reaction conditions of general applicability.3-8 With a high content of α-O-4 and β-O-4 ethereal linkages (up to 70%, Fig. 1),9,10 the efficient cleavage of the C=O bonds in lignin is a key step to promote its depolymerization and, at present, two main strategies based on oxidative or reductive conversion are being considered to obtain aromatic chemicals.

The oxidative depolymerization of lignin offers a highly efficient entry to the solubilization of lignin and a difficulty in this approach remains the isolation of pure aromatics.11,12 In contrast, mild reductive routes that would converge towards monooaromatic derivatives of the constituting monolignols (Fig. 1), by replacing C=O bonds with C-H bonds, are quite rare.13 Recently, Westwood et al isolated phenolic monomers of lignin through a two-step procedure involving first selective oxidation of Cα-OH groups and then C-O bond reduction of β-O-4 linkages with stoichiometric amounts of zinc metal.14 De Vries, Barta et al.15-17 and Luterbacher et al.18 explored over the last years an alternative approach where lignin is first depolymerized via acidolysis, in the presence of a diol or formaldehyde to stabilize reactive electrophilic intermediates and avoid re-polymerization through C-C bond formation. Subsequent hydrogenation enabled the formation of a narrow range of monomeric products. In 2015, our group reported the first catalyst (B(C6F5)3) that promotes, in the presence of hydrosilanes, the direct and efficient reductive depolymerization of a variety of softwood (resinous trees) and hardwood (leaved trees)19,20 lignins into isolable monoaromatic products.21 Depending on the wood source and the depolymerization conditions a high degree of convergence could be reached, with the formation of 1 to 4 monoaromatics. The stability of the B(C6F5)3 catalyst however seemed strongly related to the presence of lignin impurities and the nature of the solvents. More stable catalysts are therefore needed to extend the potential of a direct reductive depolymerization of lignin. Brookhart’s cationic iridium(III) pincer complex [1][B(C6F5)3] is a well-known catalyst in the hydrosilylation of C=O and C=O bonds.22-26 Notably, Brookhart et al. have shown that [1][B(C6F5)3] could cleave the C(sp3)-O bond in alkylethers to yield a silyl ether and an alkane.22 The similar chemical behavior of the Lewis acids B(C6F5)3 and [1][B(C6F5)3]27-29 led us to compare their stabilities and activities in the reduction of α-O-4 and β-O-4 linkages, both in model molecules and lignin matrices. Here, we report the first molecular metal catalyst,

![Figure 1 Convergent reductive hydrosilylation procedures for depolymerization of lignin](image)

Results and discussion

The efficiency of [1][B(C6F5)3] in the catalytic cleavage of α-O-4 and β-O-4 linkages was first evaluated on lignin model compounds. At room temperature (RT), treatment of benzyl phenyl ether 2 in the presence of 1.2 equiv. Et3SiH and with 2 mol% [1][B(C6F5)3] effectively led, after 16 h in chlorobenzene, to the selective hydrosilylation of the C(sp3)-O bond with formation of 4 as the major product (92%) and toluene and phenoxy silane 3a as minor products (<5% yield) (Scheme 1, Eq.1).

Notably, the C(sp3)-O bond in 2 was left untouched, in line with the previous findings of the Brookhart group.22 Compound 4 is a diaryl product that likely results from a Friedel-Crafts alkylation of 3a by the transient PhCH2⁺ cation (formed by deoxygenation of 2 with a silylum cation). Models 5 and 7 exhibit a β-O-4 linkage with a hydroxyl group at the α position for 5a and 5b and two hydroxyls at the α and γ positions in the
more realistic model 7. In contrast to 2, no C–O bond scission was observed when 5a/b and 7 were exposed, at RT, to 2 mol% of [1]+ with excess Et3SiH (n > 5 equiv.) and dehydrogenative silylation of the O–H groups was observed as the sole reaction (see ESI). Nevertheless, at 70 °C, 5a reacted with 3 equiv. Et3SiH and 2 mol% [1]+ to yield 3a and silyl ether 6 in 92 % yield, without recondensation (Scheme 1, Eq. 2). Similarly, the coniferyl derivative 5b was converted to 3b and 6 in 99 % yield, when 4 equiv. Et3SiH were used, and H2 and CH4 by-products were observed by GC. Using analogous conditions, 7 was successfully converted to 3a and 8 (83-99%), after hydroisilylation of the O–H and C(sp2)–O bonds (Scheme 2, Eq. 3). These reactions also proceeded with other silanes (Et3SiH2, PMHS, TMDS…) but, due to the presence of several Si-H functionalities in these reductants, analyses of the products were complicated by the number of silylated compounds (see ESI). Whereas the [1]+[B(C6F5)4] and Et3SiH system is able to reduce the C(sp2)–O bond in the α–O–4 and β–O–4 linkages, no deoxygenation of the primary (y position) and secondary (α position) alcohols was noted.

\[
\text{Scheme 1 Reductive cleavage of simple α-O-4 and β-O-4 model 2 (eq. (1)) and 5a and 5b (eq. (2)) using Et3SiH and the catalyst [1]+[B(C6F5)4].}
\]

These observations are consistent with previous results on C–O bond hydrosilylation of ethers\textsuperscript{22} and the alcoholysis of hydroxilanes by cationic iridium complexes.\textsuperscript{30} The selective formation of 6 from 5a/b, instead of the expected silylated 1-phenylethanol (PhCH(OSiEt\textsubscript{3})Me), likely results from a semi-pinacol rearrangement with migration of a phenyl group, as observed with the B(C6F5)\textsubscript{3}]+Et3SiH system.\textsuperscript{31} A similar rearrangement accounts for the formation of 8. The coniferyl alcohol derivative 9, which differs from 7 by the presence of additional methoxy substituents on the aromatic rings, is a more realistic structural model of lignin.\textsuperscript{22} Hydroisilylation of 9 with an excess Et3SiH (8 equiv.) and 2 mol% [1]+[B(C6F5)4] afforded the tris-silylated primary alcohol 10G (>80% yield), along with 3b (Scheme 2, Eq. 4). Further reduction of the C–O–Si bond in 10G, to yield the propyl derivative CH\textsubscript{3}-(CH\textsubscript{2})\textsubscript{2}-(CH\textsubscript{3})\textsubscript{2}(OSiEt\textsubscript{3})\textsubscript{3}, only proceeded under harsher conditions (90°C for 5 days, see ESI). Interestingly, the formation of 10G shows that the reduction of 9 differs from the reduction of 5 and 7 and occurs without any semi-pinacol rearrangement, thereby exemplifying the importance of the aromatic substituents on the reactivity of lignin model compounds. To better understand this change in reactivity, DFT calculations were performed on model structures of 5 and 9 (see Scheme 3 and ESI). Upon activation of silylated 5 with the R3Si\textsuperscript{+} cation (generated by the electrophilic activation of the hydrosilane with the Lewis acidic catalyst), the C–O bond cleavage in 5a is kinetically favored. Yet, the formation of the corresponding cation 6a\textsuperscript{+}, bearing a positive charge on C3a, is endergonic (ΔG=+4.1 kcal/mol) and 5a\textsuperscript{+} rather rearranges to the more stable 6a\textsuperscript{+} intermediate by semi-pinacol rearrangement (ΔG=−10.4 kcal/mol). The latter cation then accumulates prior to its reduction to 6 (and 3a). In contrast, when methoxy substituents are introduced on the phenyl ring (as in 9), the stability of MeO-6a\textsuperscript{+} increases and its formation becomes exergonic (ΔG=−2.7 kcal/mol) and kinetically favored, thereby leading to a primary silyl ether. This behavior directly results from the enhanced stability of a positive charge on the C3a center provided by the electron donating methoxy substituents in G and S units.

Results depicted in Eqs. 1–4 provide interesting trends for comparing the [1]+[B(C6F5)4] and B(C6F5)\textsubscript{3} catalysts in the hydrosilylation of lignin models (reaction conditions: 2 mol% catalysts, Et3SiH as reductant, 16 h). The two catalysts share similar mechanistic schemes, leaving C(sp2)–O bonds untouched and promoting semi-pinacol rearrangements in 5 and 7. While B(C6F5)\textsubscript{3} is active at RT, heating to 70 °C is required for [1]+[B(C6F5)4]. The latter catalyst overall exhibits an enhanced chemoselectivity and it is able to reduce selectively the C3–O and C4–O bonds in 9 with an excess hydrosilane, whereas C2–O, C3–O and C4–O bond cleavages are competing with the B(C6F5)\textsubscript{3}/Et3SiH system, leading to over-reduction.

Having in hand an efficient catalytic system for the reductive cleavage of α–O–4 and β–O–4 linkages, the depolymerization of natural lignin was then attempted, with the aim to overcome the structural diversity and complexity of this biopolymer and access to narrow range of products, in a single step. Lignin, extracted from industrial pine with a Formacell process\textsuperscript{7} was exposed to a chlorobenzene solution of [1]+[B(C6F5)4] (25 wt% relative to lignin weight, corresponding to ca. 3.5 mol% per aromatic unit) and a slight
excess of Et₃SiH (273 wt%, ca. 5 Si–H functionalities per aromatic unit). Within 24 h at 90 °C, the complete dissolution of the lignin matrix was observed and 10G was obtained as the sole monoaromatic product. Compound 10G was isolated in 99 wt% yield as colorless oil after purification by chromatography (Scheme 4). The expected diaromatic compounds (which would represent a significant amount of the hydrosilylated products = 140 wt% for industrial pine lignin) could not be isolated because they are retained on the chromatographic column. The mass of 10G corresponds to a molar yield of 38 % in aromatic compounds (see ESI). The formation of 10G demonstrates the successful transposition of the [1]B(C₆F₅)₃/ Et₃SiH hydrosilylation system to natural lignin and exemplifies the ability of [1]⁺ to convert highly functional substrates.

Optimization of the depolymerization conditions, by extending the reaction time to 48 h in the presence of a low catalyst charge of 10 wt% (ca. 1.3 mol% per aromatic unit) with 300 wt% Et₃SiH at 70 °C (Table 1), enabled the formation of 10G in 120 wt% yield, corresponding to a molar yield of 46 %.

<table>
<thead>
<tr>
<th>Scheme 4</th>
<th>Reductive depolymerization of Formacell lignin (derived from industrial pine) to 10G, using [1]B(C₆F₅)₃/ Et₃SiH.</th>
</tr>
</thead>
<tbody>
<tr>
<td>Entry</td>
<td>[1]B(C₆F₅)₃ (wt%)</td>
</tr>
<tr>
<td>1</td>
<td>50</td>
</tr>
<tr>
<td>2</td>
<td>20</td>
</tr>
<tr>
<td>3</td>
<td>20</td>
</tr>
<tr>
<td>4</td>
<td>10</td>
</tr>
<tr>
<td>5</td>
<td>5</td>
</tr>
</tbody>
</table>

Table 1 Optimization of [1]B(C₆F₅)₃/ Et₃SiH system on industrial pine lignin.

Interestingly, in presence of 5 wt% [1]B(C₆F₅)₃, no depolymerization products could be identified by GC-MS and NMR spectroscopies, although the lignin matrix was completely solubilized, from the hydrosilylation of the hydroxyl groups (Entry 5, Table 1).

This would suggest that a minimum quantity of catalyst is required because common impurities in lignin, such as residual carbohydrates and lipids, are known to deactivate and poison catalyst systems. Replacing industrial pine with other softwood varieties, namely Pacific red cedar, Lebanese cedar and Norway spruce, Formacell lignins were deconstructed to 10G, in 11, 69 and 81 wt% yield, respectively, with molar yields ranging 4 to 31 % (Scheme 5, Eq. 7).

As depicted in Scheme 5, the reactions are highly selective leading to 10G as the major monoaromatic product. The other diaromatic compounds were not detected in GC-MS because of their excessive weights. The depolymerization of hardwood lignins was carried out next. In addition to contain a greater proportion of β-O-4 cleavable linkages than softwood, these lignins present both sinapyl (S) and coniferyl (G) residues.

Scheme 5 Depolymerization of softwood lignins with [1]B(C₆F₅)₃/ Et₃SiH to 10G (Eq. 7). GC-MS chromatograms of the crude reactional mixture for a variety of softwood lignins: 1) CHCl₃ (solvent added for GM-MS analyses), 2) C₆H₅Cl (3), 3) Et₃SiOH, 4) Et₃SiH, 5) Et₃SiOSiEt₃.

They are thus more oxidized and require forcing conditions with a larger charge of hydrosilane. As a consequence, the hydrosilylation of lignin samples obtained from black poplar, evergreen oak, hybrid plane and common beech was achieved at 70 °C in chlorobenzene, with 20 wt% catalyst and 364 wt% Et₃SiH (i.e. ca. 6.3 molar equivalents per aromatic unit) (Scheme 6, Eq. 8).

GC chromatograms recorded after 48 h showed 10S and 10G to be the main monoaromatic products. In contrast to the resinous counterparts, they also evidenced a number of minor and undetermined side products (see ESI). For the hardwood lignins, the quantities of 10S are larger than that of 10G, with yields varying within 80-130 wt% and 20-42 wt%, respectively, in line with their relative proportions in the lignin matrices (e.g. ratio 10S/10G: 70/30, see ESI). Here again, the mean molar yields depend on the wood type and are quite high ranging 36 % for black poplar to 58 % for evergreen oak (Eq. 8).

While the reductive depolymerization of lignin could be achieved within hours at RT when catalysed with B(C₆F₅)₃, Brookhart’s catalyst appears less reactive requiring heating to 70 °C. To compare precisely the two catalytic systems, depolymerization experiments of pine lignin were thus carried out with the same molar quantities (2.9 µmol) of catalyst, namely 10 wt% [1]B(C₆F₅)₃ and 3 wt% B(C₆F₅)₃.

Under these conditions, as summarized in Scheme 7, the boron catalyst proved inefficient at RT or 70°C, both in dichloromethane and chlorobenzene, and no trace of 10G could be detected by GC-MS. The lack of activity of B(C₆F₅)₃ certainly results from its degradation or poisoning. This also reveals the enhanced robustness of the iridium catalyst which displays a
TON of 28 considering the molar yield of monoaromatic products with respect to the catalyst loading (see ESI).\textsuperscript{5}

Isolation of phenolic compounds instead of the silylated species was important as the former are valuable synthons in chemical synthesis and industry. Hydrolysis of compounds 10S and 10G with hydrated tetrabutyl ammonium fluoride afforded the corresponding propylhydroxycathecols 10S’ and 10G’ in 94 and 84 % yield, respectively. 10S’ and 10G’ were thus isolated, respectively, in 55 % molar yield from evergreen oak lignin (30 wt%/wt lignin) and 36 % molar yield from industrial pine lignin (26.5 wt%/wt lignin) (scheme 8). These are the highest yields in isolated mono-aromatic compounds ever obtained from lignin under homogeneous conditions.\textsuperscript{7,8} The cost of the catalysts as well as the release of wastes (siloxanes, fluorosilanes) are however drawbacks of the present method. Current efforts are being devoted to improve the sustainability of hydrosilylation chemistry, for example through the use of silylformates.\textsuperscript{33} In addition, formic acid would be an attractive reductant in lignin depolymerization.

Another key difference between the two catalysts [1]\textsuperscript{1} and B(C\textsubscript{6}F\textsubscript{5})\textsubscript{3} lies on their distinct chemoselectivity. While 10G and 10S are obtained selectively from the treatment of softwood and hardwood lignins with [1]\textsuperscript{1} and an excess hydrosilane, respectively, over-reduced derivative with the propyl chain was only obtained with the use of B(C\textsubscript{6}F\textsubscript{5})\textsubscript{3}.\textsuperscript{19}

The efficiency of lignin depolymerization with [1][B(C\textsubscript{6}F\textsubscript{5})\textsubscript{3}] was quantified as the ratio of the experimental yield in monoaromatic products on the theoretical yield (see ESI). It is different from the molar yield because not all the interunit linkages in lignin can be reduced. This calculation required the knowledge of the mean molecular weight of Formacell lignin (determined by SEC analyses) and the content in cleavable linkages (e.g. β-O-4 content).\textsuperscript{21} The amount of β-O-4 linkages was evaluated from the HSQC NMR spectra of the lignin samples by two classical approaches, i) when quantified with respect to aromatics units or ii) when expressed as a percentage of the total side chain linkages (See details in ESI). The β-O-4 contents vary in the range of 42-48 per 100 aromatics and 65-70% of the interunit linkages in hardwood lignin (Table 2). While the former approach (i) fails and provides aberrant results (see ESI), the second approach was used, affording efficiencies in the range of 42 to 100 % (Table 2). Overall, softwood and hardwood lignins exhibit similar efficiencies.

<table>
<thead>
<tr>
<th>Lignin source</th>
<th>β-O-4’ content</th>
<th>Molar yield in 10G or 10G + 10S (%)</th>
<th>Theoretical yield (%)</th>
<th>Eff(\textsuperscript{1})</th>
<th>Yield in phenolic compounds (10G’ or 10S’)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Industrial Pine</td>
<td>63 %\textsuperscript{(n)} (35.1)\textsuperscript{(n)}</td>
<td>46 %</td>
<td>46 %</td>
<td>100 %</td>
<td>38 %</td>
</tr>
<tr>
<td>Norway Spruce</td>
<td>46 %</td>
<td>(20.2)</td>
<td>31 %</td>
<td>35 %</td>
<td>89 %</td>
</tr>
<tr>
<td>Lebanese Cedar</td>
<td>59 %</td>
<td>(38.9)</td>
<td>27 %</td>
<td>46 %</td>
<td>58 %</td>
</tr>
<tr>
<td>Pacific Red Cedar</td>
<td>20 %</td>
<td>(21)</td>
<td>4 %</td>
<td>10 %</td>
<td>42 %</td>
</tr>
<tr>
<td>Evergreen Oak</td>
<td>70 %</td>
<td>(48.1)</td>
<td>58 %</td>
<td>60 %</td>
<td>96 %</td>
</tr>
<tr>
<td>Hybrid Plane</td>
<td>65 %</td>
<td>(42.5)</td>
<td>47 %</td>
<td>60 %</td>
<td>79 %</td>
</tr>
<tr>
<td>Common Beech</td>
<td>65 %</td>
<td>(43.7)</td>
<td>46 %</td>
<td>60 %</td>
<td>77 %</td>
</tr>
<tr>
<td>Black Poplar</td>
<td>76 %</td>
<td>(47.2)</td>
<td>36 %</td>
<td>60 %</td>
<td>n.d.</td>
</tr>
</tbody>
</table>
Conclusions

In summary, we have reported herein the first use of the Brookhart’s iridium complex as a catalyst in the reductive cleavage of lignin model molecules and softwood and hardwood lignins. Combined with hydroxilanes, \([1]B(C_6F_5)₄\) favors selective splitting of the \(C(sp^3)\)–O bonds in alkyl ethers and secondary silyl ether to provide silylated monoaromatic chemicals, with coniferyl \(G\) and sinapyl \(S\) residues, that can be isolated with high yields in a pure form. Although \([1]B(C_6F_5)₄\) displays a lower catalytic activity than \(B(C_6F_5)₃\), this complex exhibits a greater stability and selectivity and represent a rare example of a catalyst able to depolymerize lignin to isolable monoaromatics.

Acknowledgements

For financial support of this work, we acknowledge the CEA, CNRS, CHARMMAT Laboratory of Excellence and the European Research Council (ERC Starting Grant Agreement n.336467). T.C. thanks the Fondation Louis D. – Institut de France for its support. We thank CINES for the allowance of computer time (Project No. c2017086494). The authors thank Dr Patrick Berthaud for his assistance with NMR experiments.

Notes and references

† We have previously reported from steric exclusion chromatography (ref. 19) that lignin obtained from the formacell organosolv process consisted in oligomers featuring a mean sequence of 5 aromatic units, with mainly \(S\) and \(G\) aromatic units (Scheme 1) for hardwood and essentially \(G\) units for softwood.

‡ 1.3 mol% is found with \([1]B(C_6F_5)₄\) (1351 g.mol\(^{-1}\)) and a mean molecular weight for industrial pine lignin of 1099 g mol\(^{-1}\).

§ If one considers that the production of 10G from softwood lignin requires 6 C=O and O–H bond silylation events, then the TON is 170 per aromatic units (see ESI).

Uncategorized References


